

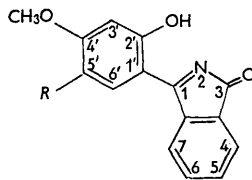
During the course of the preliminary investigation it was noted in a periodical review of the literature that Pepinsky, Jona & Shirane (1956) had observed the appearance at 176 °K. of a peak in the dielectric constant of methyl ammonium alum, the occurrence of the peak being accompanied by a sharp maximum in the loss factor; this effect is presumably similar to that reported by Griffiths & Powell (1952).

Acta Cryst. (1961). **14**, 893

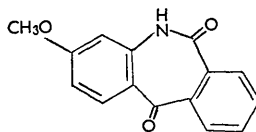
An X-ray crystallographic study of 1-(5'-bromo-2'-hydroxy-4'-methoxyphenyl)-3-oxo-isoindolo-1-en. By NOEL E. TAYLOR, *Department of Physics, University of Cape Town, South Africa*

(Received 3 January 1961)

Lamchen (in preparation) has described the synthesis of a red crystalline compound which he has formulated as 1-(2'-hydroxy-4'-methoxyphenyl)-3-oxo-isoindolo-1-en (Ia). An alternative structure (II), with the same empirical formula ($C_{15}H_{11}NO_3$), could be suggested for this compound and the present study was undertaken to decide between these alternatives. To enable a direct structure determination by the 'heavy-atom method', a bromo-derivative was prepared in which the bromine atom was substituted on to the phenyl group at one of the positions 3', 5'- or 6'; on chemical grounds, the 5'-position (structure Ib) seemed most probable and has been verified by the X-ray analysis.



(Ia): $R = H$
(Ib): $R = Br$



(II)

The crystals grow from a chloroform-petroleum-ether solution in the form of very fine, deep-red needles which are frequently twinned. X-ray oscillation and Weissenberg photographs showed them to be triclinic, with one very short unit-cell dimension parallel to the needle axis.

Crystal data

$$C_{15}H_{10}NO_3Br, M = 332.2,$$

$$a = 3.96 \pm 0.01, b = 11.91 \pm 0.03, c = 13.53 \pm 0.04 \text{ \AA},$$

$$(\lambda(\text{Cu } K\alpha) = 1.5418 \text{ \AA})$$

$$\alpha = 100^\circ 46' \pm 20', \beta = 91^\circ 40' \pm 20', \gamma = 91^\circ 20' \pm 20',$$

$$U = 625.6 \text{ \AA}^3, D_m = 1.76 \text{ g.cm.}^{-3}, Z = 2, D_c = 1.763 \text{ g.cm.}^{-3},$$

$$F(000) = 332.$$

The $0kl$ intensity data were measured by visual estimation, from a series of multiple-film Weissenberg photographs. Of the 417 non-equivalent reflections within the $\text{Cu } K\alpha$ limit, 339 were recorded. A statistical survey of the intensity data gave an $N(z)$ distribution approximating more closely to the theoretical curve for a non-centrosymmetric projection (Howells, Phillips & Rogers, 1950), suggesting the space group $P1$. However, when allowance is made for the predominant heavy bromine atoms, as

References

- GRIFFITHS, J. H. E. & POWELL, J. A. (1952). *Proc. Phys. Soc. A*, **65**, 289.
LIPSON, H. (1935a). *Phil. Mag.* **19**, 887.
LIPSON, H. (1935b). *Proc. Roy. Soc. A*, **151**, 347.
PEPINSKY, R., JONA, F. & SHIRANE, G. (1956). *Phys. Rev.* **102**, 1181.

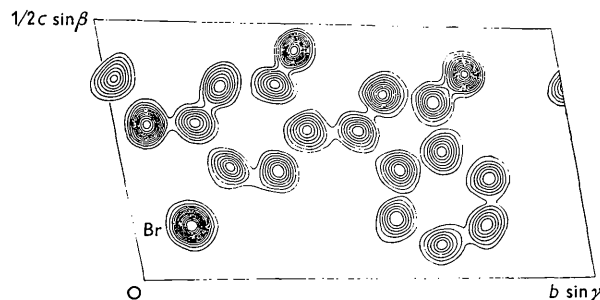


Fig. 1. The electron-density projection normal to the a -axis. The first contour is at 2 e.\AA^{-2} and the contour interval is 1 e.\AA^{-2} , except over the bromine atom where the interval is 5 e.\AA^{-2} .

suggested by Sim (1958), the results seem to favour the centrosymmetric space group $P\bar{1}$.

(y, z) parameters for all the atoms in the molecule were derived from an electron-density projection calculated with the phases of the bromine contributions to the structure factors. A set of $0kl$ structure factors based on these atomic positions showed satisfactory agreement with the observed structure amplitudes (the reliability factor $R = \sum |F_o - F_c| / \sum |F_o| = 0.17$). The final electron-density distribution is shown in Fig. 1 where a clearly resolved projection of the atomic maxima corresponding to a molecule of structure Ib can be seen. The average peak height over the carbon atoms is 8.8 e.\AA^{-2} and over the oxygen atoms is 12.3 e.\AA^{-2} , while the peak height of the one nitrogen atom is 11.0 e.\AA^{-2} .

The present two-dimensional analysis confirms the proposed structure (Ib). Packing considerations suggest that the molecules are linked by a system of $\text{O-H} \cdots \text{N}$ bonds approximately 2.8 \AA in length; a complete analysis is required to investigate this hydrogen bonding and to provide details of the molecular geometry.

The author wishes to thank Prof. R. W. James, F.R.S., for his interest in this work. Financial assistance from the South African Council for Scientific and Industrial Research is gratefully acknowledged.

References

- HOWELLS, E. R., PHILLIPS, D. C. & ROGERS, D. (1950). *Acta Cryst.* **3**, 210.
SIM, G. A. (1958). *Acta Cryst.* **11**, 123.